

## Erratum

# Test of variational transition state theory with multidimensional tunneling contributions against experimental kinetic isotope effects for the $\text{CH}_n\text{D}_{4-n} + \text{OH} \rightarrow \text{P}$ ( $n = 0, 4$ ) reactions

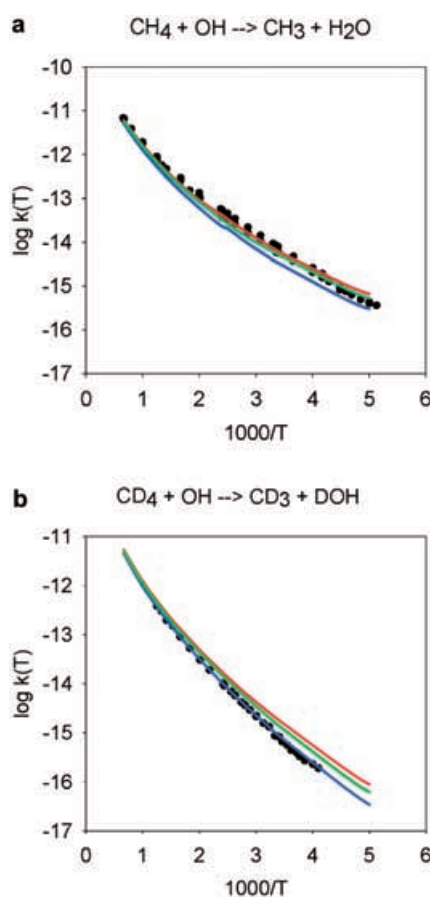
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Due to a technical error Fig. 1 (a and b) was published in black and white. The figure in colour is shown below.



**Fig. 1a, b.** Arrhenius plots for the experimental and calculated rate constants of the reactions: **a**  $\text{CH}_4 + \text{OH} \rightarrow \text{CH}_3 + \text{H}_2\text{O}$ ; **b**  $\text{CD}_4 + \text{OH} \rightarrow \text{CD}_3 + \text{DOH}$ . Rate constants are in  $\text{cm}^3 \text{molecule}^{-1} \text{s}^{-1}$  and temperatures in Kelvin. Experimental data from Ref. [1] (circles); MCCM-CCSD(T)-1sc (red); MCCM-CCSD(T)-2sc (green); MCCM-CCSD(T)-2m (blue)